

Krebs Cycle (KC)

- Also known as TCA cycle, or citric acid cycle
- Reactions of KC occur in mitochondrial matrix
- Common final degradative pathway for breakdown of monomers of CHO, fat and protein to CO_2 and H_2O
 - Electrons removed from acetyl groups & attached to NAD^+ & FAD
 - Small amount of ATP produced from substrate level phosphorylation
- KC also provides intermediates for anabolic functions (eg. Gluconeogenesis)
- Anaerobic metabolism (glycolysis & fermentation) only releases 7% of the energy in glucose)
- Glycolysis is used for **rapid** ATP production :Rate of ATP formation in anaerobic is 100 times > ATP production by oxidative phosphorylation.

Some General Features of TCA Cycle

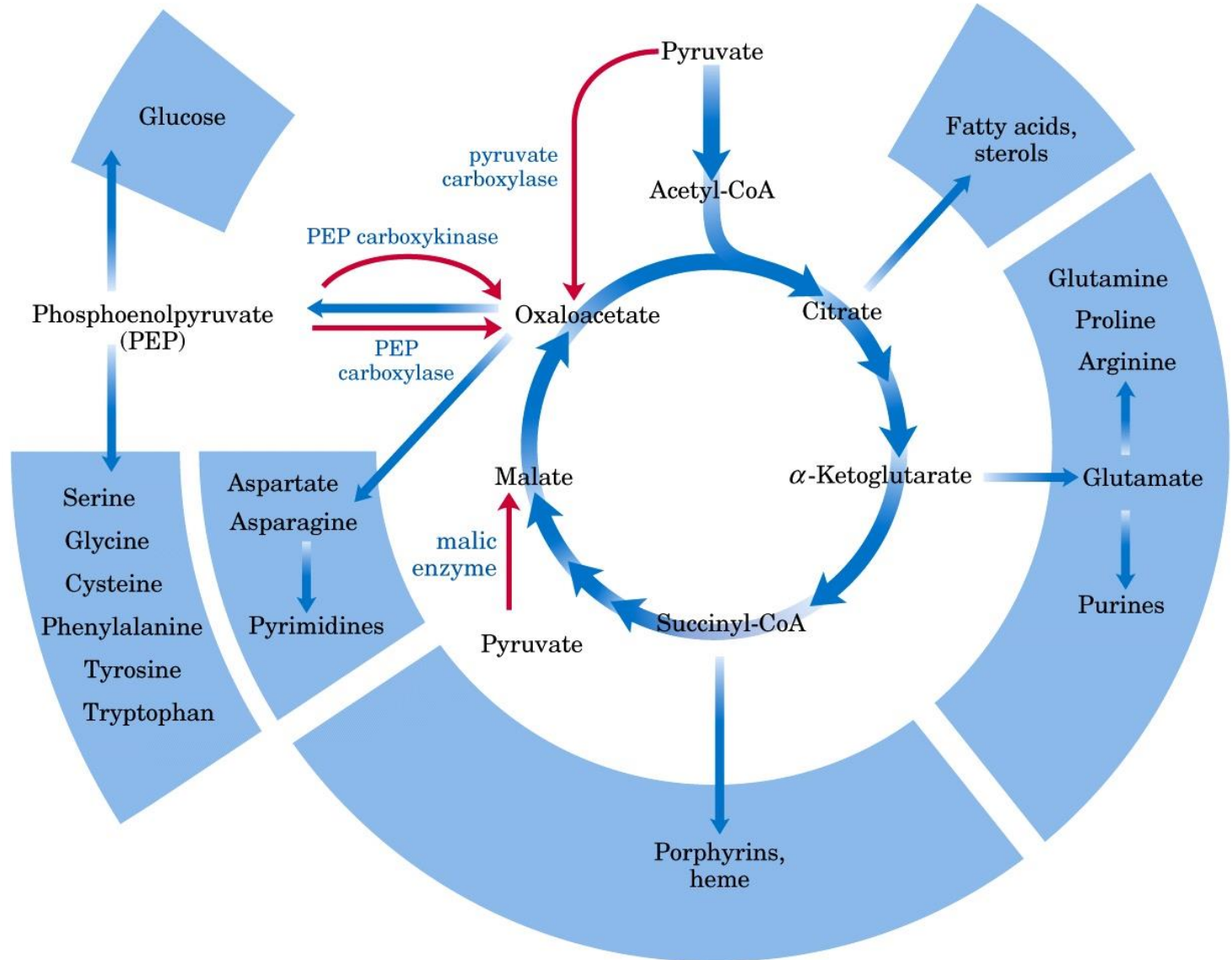
- The enzymes of TCA are found in the mitochondrial matrix.
- **Catabolic role** Amino acids, fats sugars enter the TCA cycle to produce energy.
- **Anabolic role** TCA cycle provides starting material for fats and amino acids. Note: carbohydrates cannot be synthesized from acetyl-CoA by humans. Pyruvate---Acetyl CoA is one way!
- In contrast to glycolysis, none of the intermediates are phosphorylated but all are either di- or tricarboxylic acids.

Amphibolic Nature of TCA Cycle:

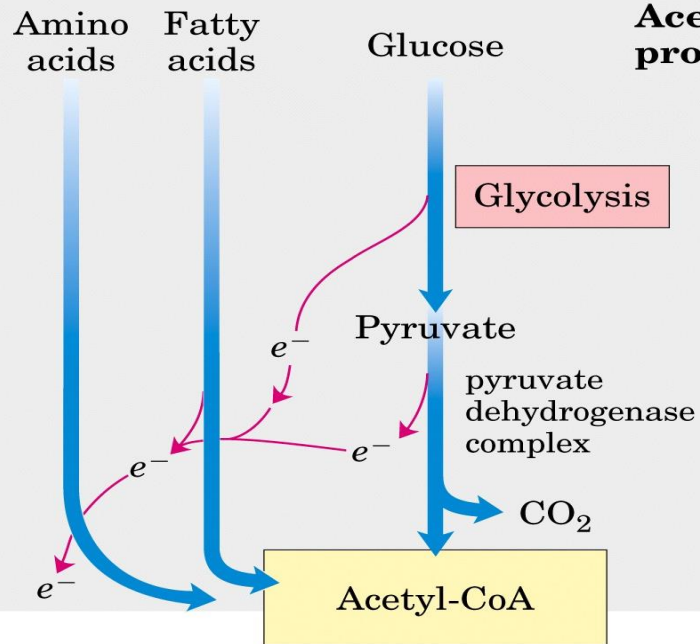
Some compounds feed in and some are removed for other uses

Anabolic

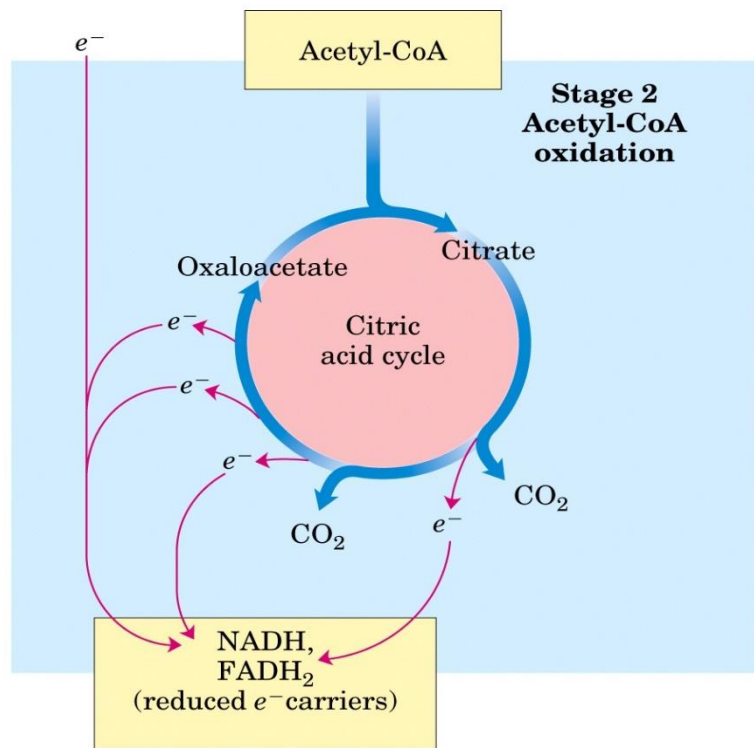
Catabolic



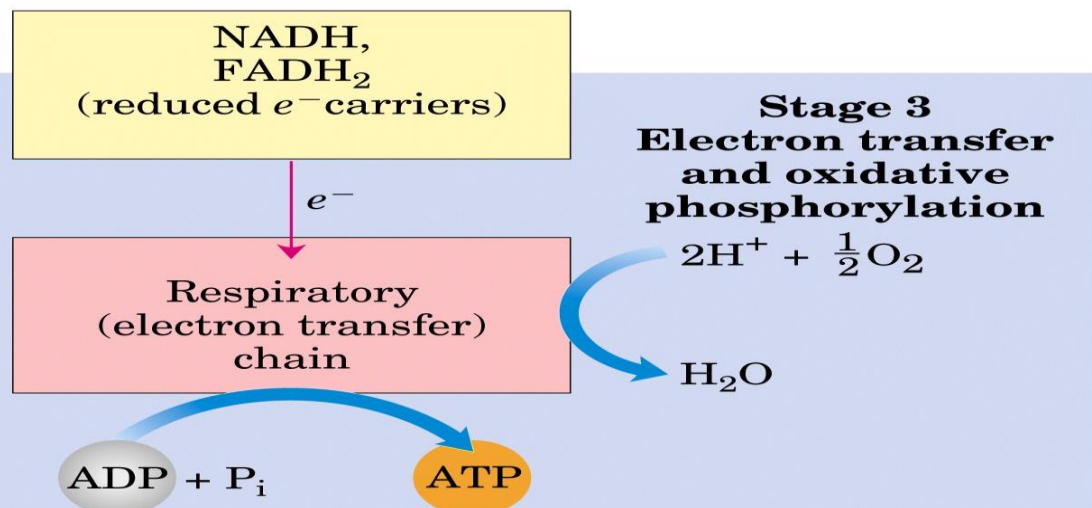
Stage 1 Acetyl-CoA production

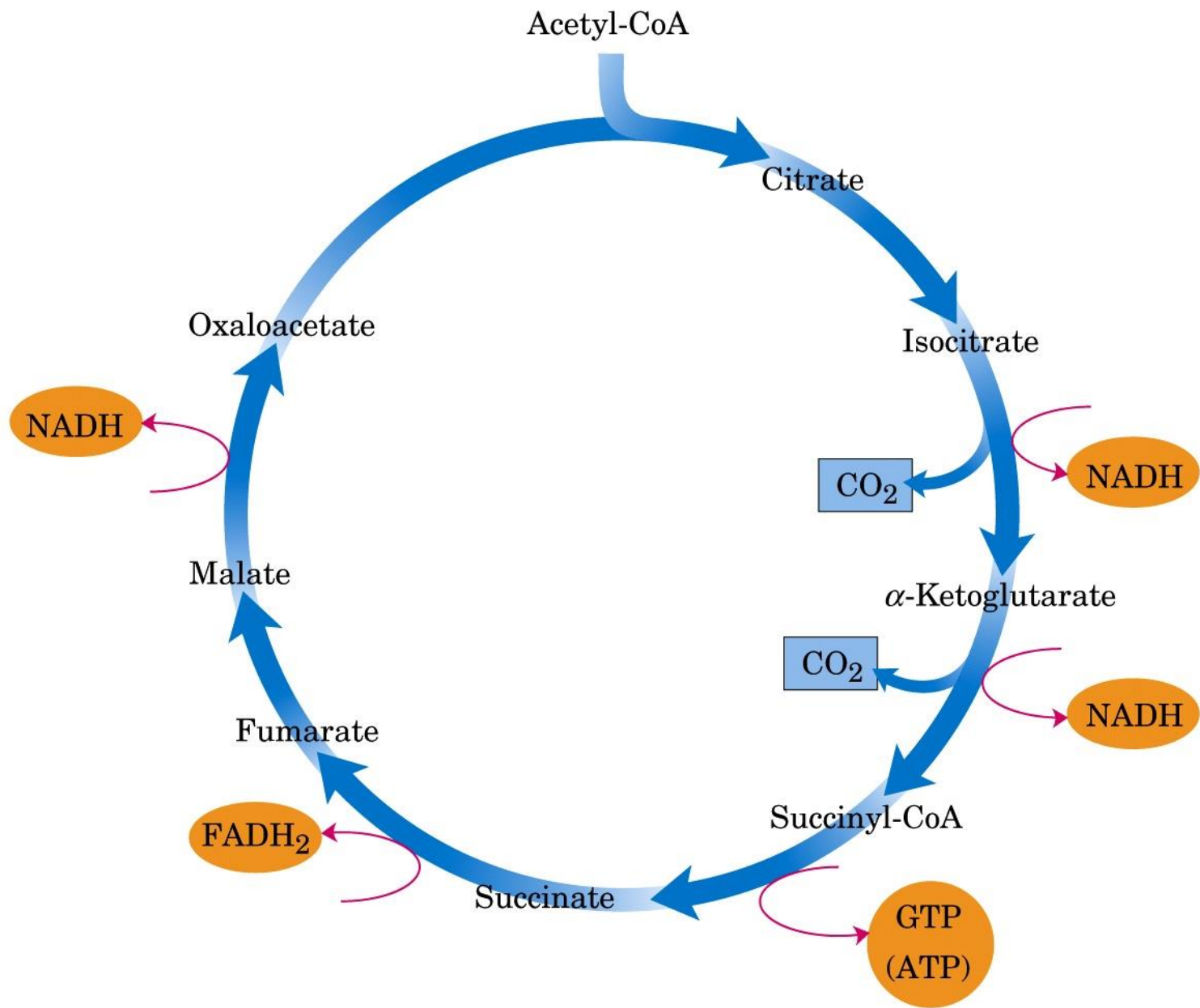


Stage 2 Acetyl-CoA oxidation



Stage 3 Electron transfer and oxidative phosphorylation



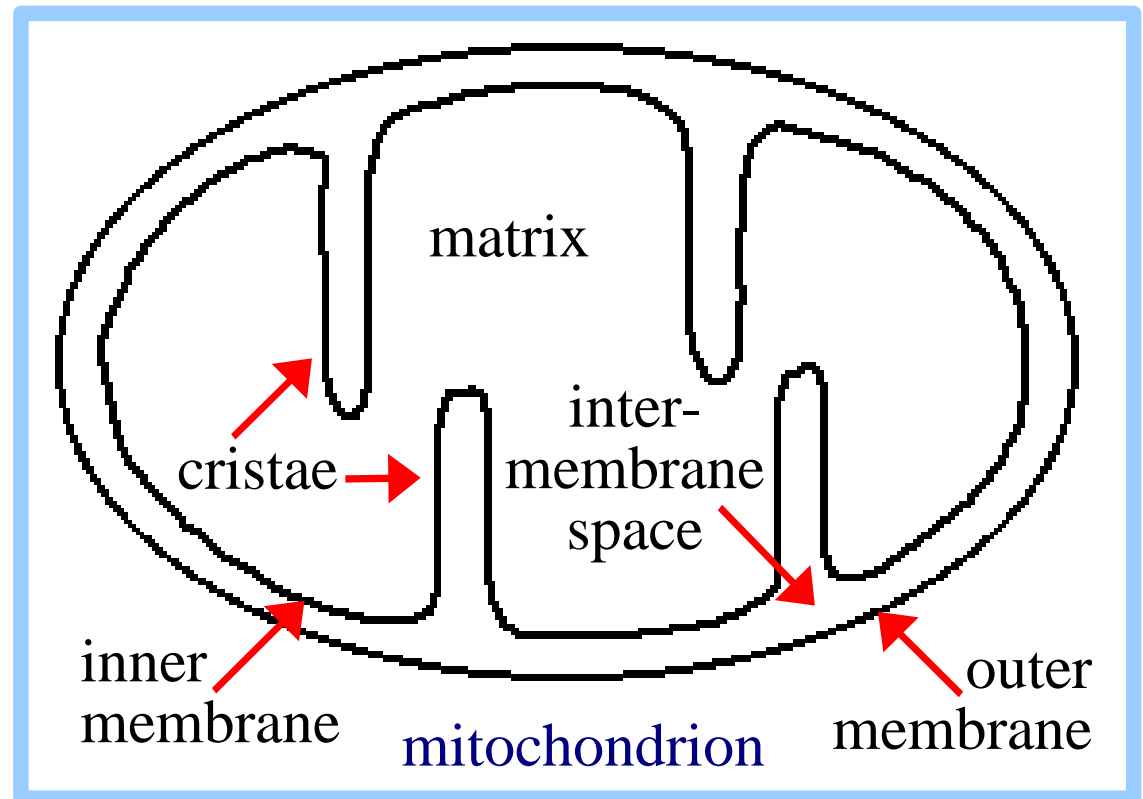


Pyruvate transport and mitochondrial structure:

The mitochondria enclosed by a double mb. All of the glycolytic enzymes are found in the cytosol of the cell. Charged molecules such as pyruvate must be transported in and out of the mitochondria.

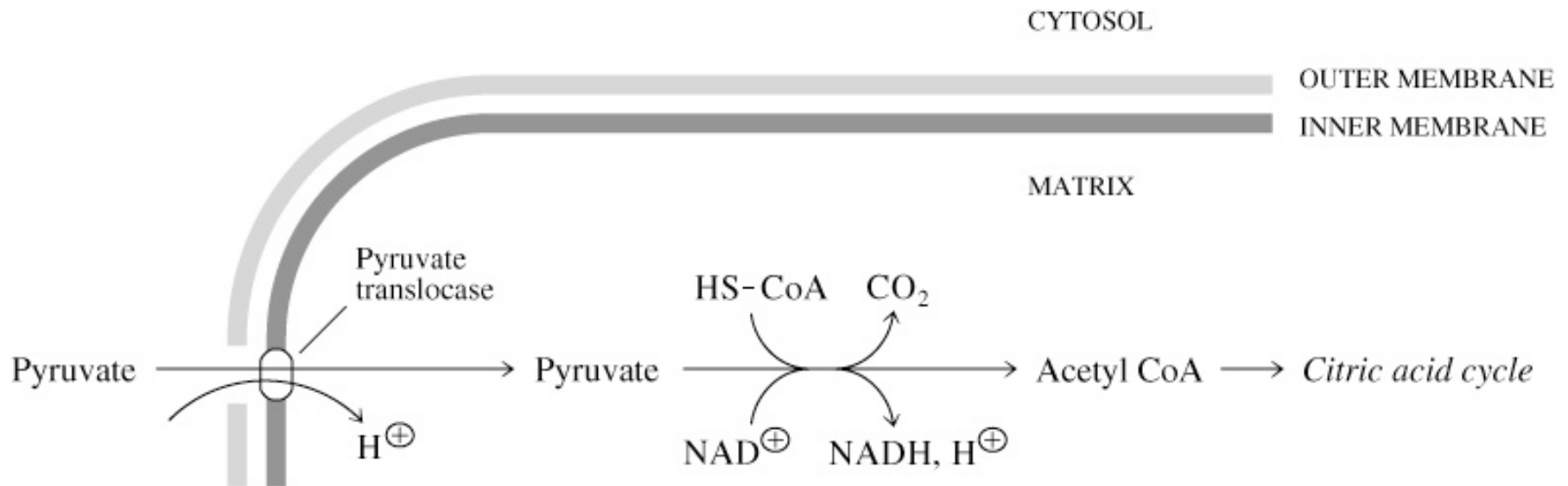
Small charged molecules with M.wt. < 10000 can freely diffuse through outer mitochondrial mb. through aqueous channels called porins.

A transport protein called pyruvate translocase specifically transports pyruvate through the inner mitochondrial mb. into the Mitochondrial matrix.



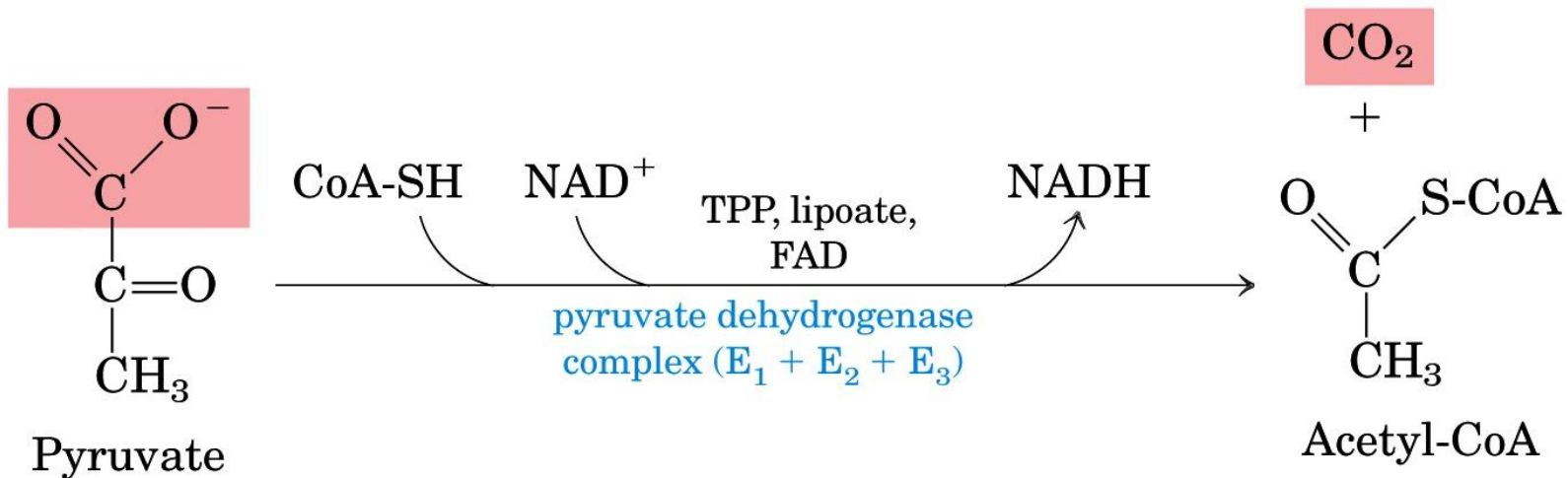
Entry of Pyruvate into the Mitochondrion

Pyruvate translocase transports pyruvate into the mitochondria in symport with H^+



Pyruvate \rightarrow Acetyl CoA

- Pyruvate produced in cytosol and transported into mitochondria
- Cannot directly enter KC
 - First converted to acetyl CoA by pyruvate dehydrogenase complex
- Oxidative decarboxylation of pyruvate

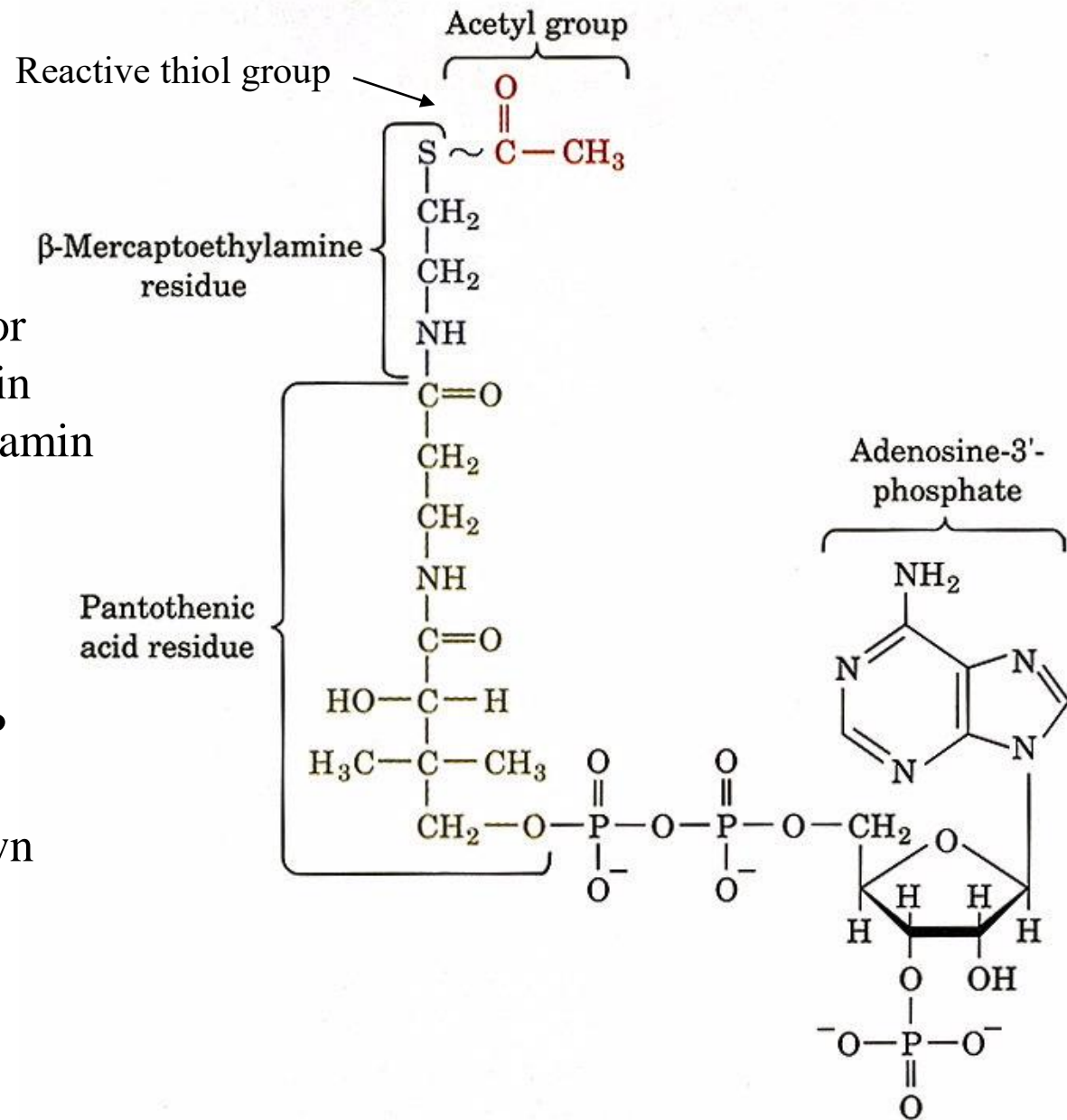


Coenzyme A

-Coenzyme: an organic cofactor required for the action of certain enzymes often containing a vitamin as a building block.

-“high energy” compound
- Hydrolysis of thioester bond more exergonic than ATP

-acetylCoA common breakdown product of CHO, FAs, AAs



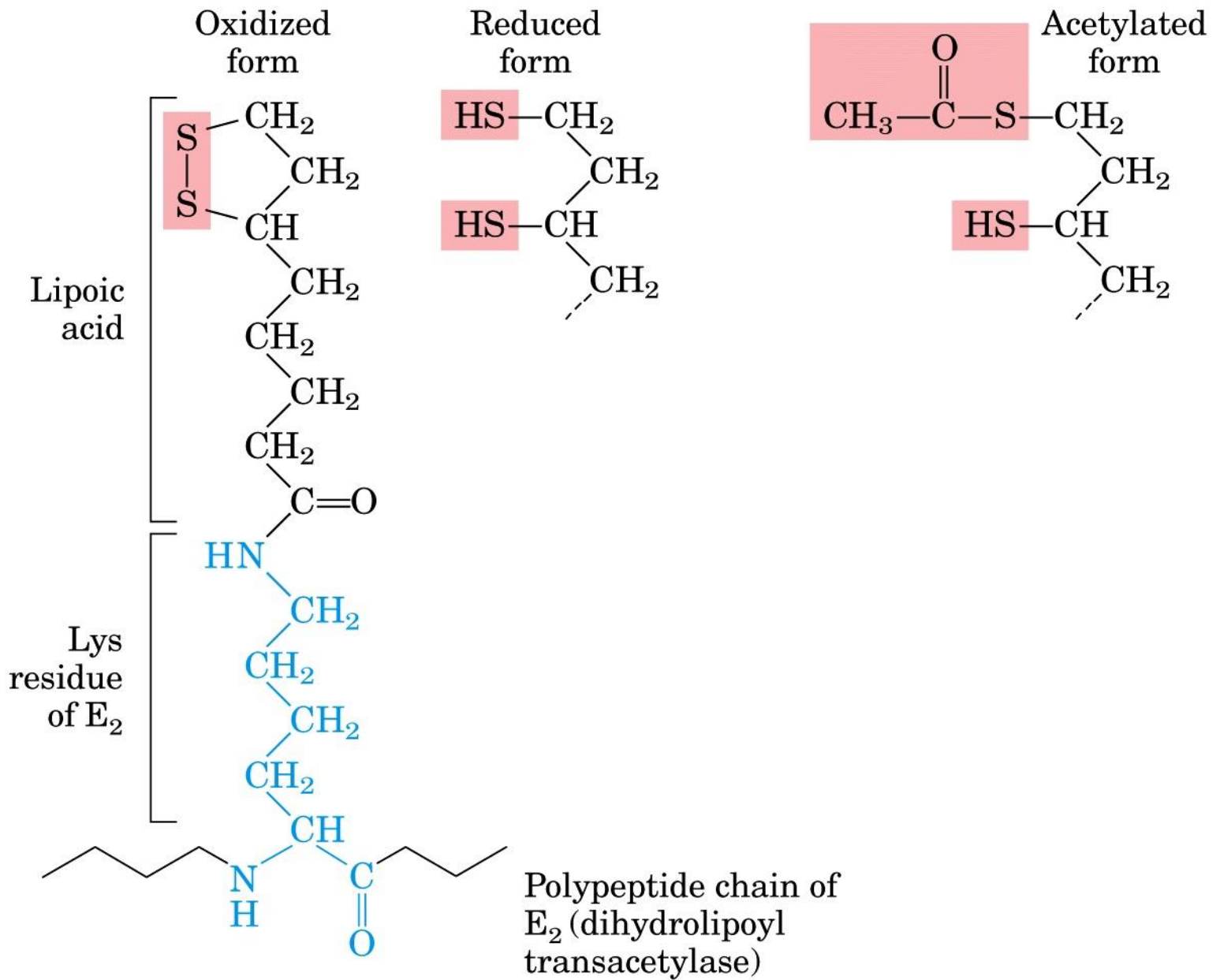
Acetyl-coenzyme A (acetyl-CoA)

Pyruvate Dehydrogenase Complex (PDH)

Enzyme	Abbreviated	Prosthetic group	Function
Pyruvate dehydrogenase	E1	Thiamine Pyrophosphate TPP*	Decarboxylation and aldehyde group transfer
Dihydrolipoyl transacetylase	E2	Lipoamide, Co-A	Carrier of hydrogens or acetyl group
Dihydrolipoyl dehydrogenase	E3	FAD**, NAD	Electron carriers

***TPP** is a derivative of thiamine (vitamin B1). Nutritional deficiency of thiamine leads to the disease **beriberi**. Beriberi affects especially the **brain**, because TPP is required for CHO metabolism, & the brain depends on glucose metabolism for energy.
Limb weakness, heart disease (green vegetables, fruit, fresh meat)

** **Flavin Adenine Dinucleotide** is derived from the vitamin riboflavin



Enzymes and Reactions of the Citric Acid Cycle

- Multi-step catalyst
 - Citrate synthase
 - Aconitase
 - Isocitrate Dehydrogenase
 - alpha-Ketoglutarate Dehydrogenase
 - Succinyl-CoA Synthetase
 - Succinate Dehydrogenase
 - Fumarase
 - Malate Dehydrogenase

Formation of citrate

Oxaloacetate condenses with acetyl CoA to form Citrate

Non-equilibrium (irreversible) reaction catalysed by citrate synthase

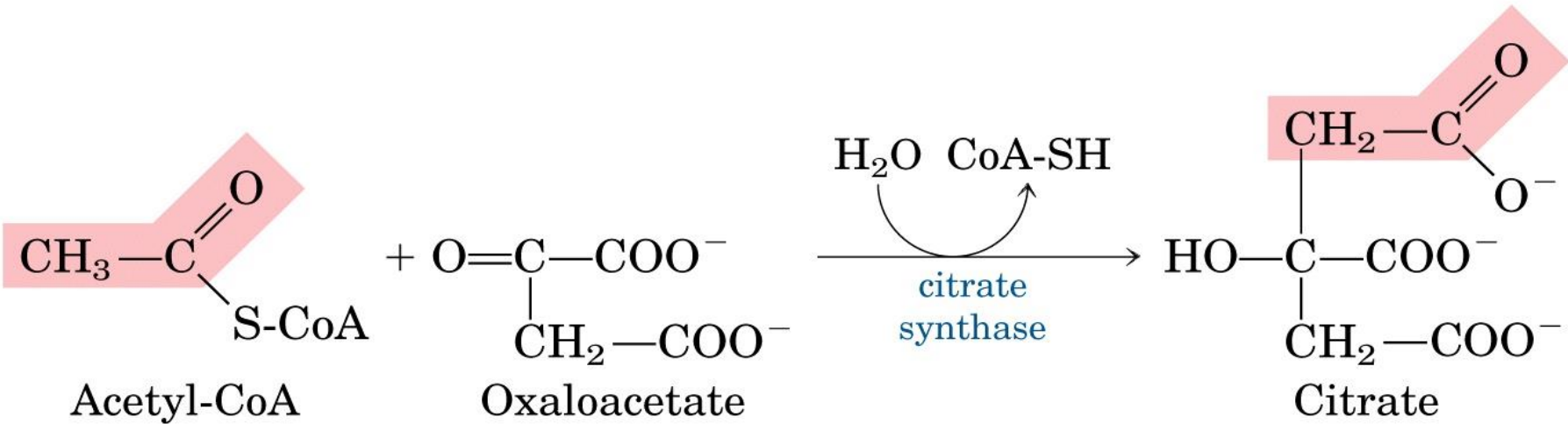
Inhibited by:

ATP

NADH

Succinyl-CoA

Citrate - competitive inhibitor of oxaloacetate

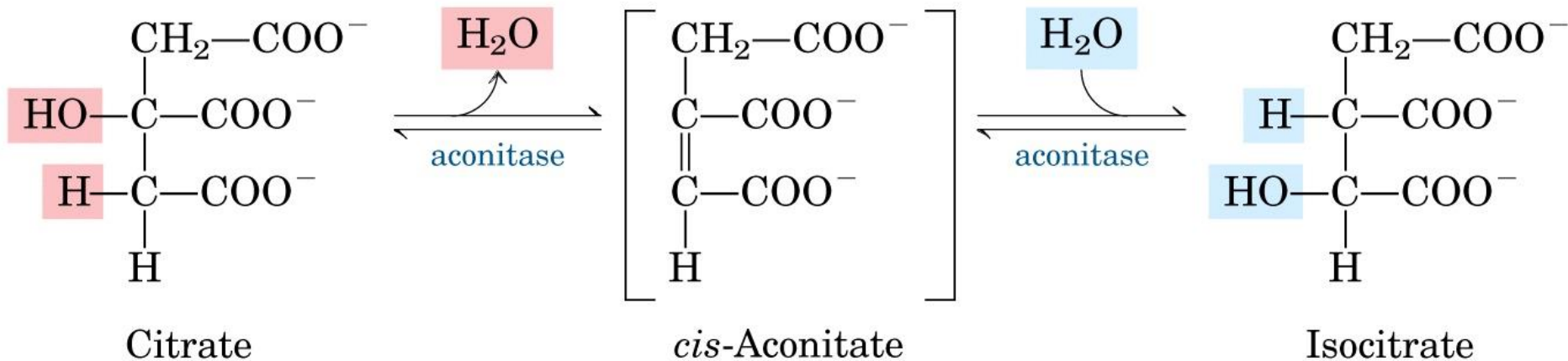


Citrate ↔ isocitrate

Citrate isomerised to isocitrate in two reactions (dehydration and hydration)

Equilibrium reactions catalysed by aconitase

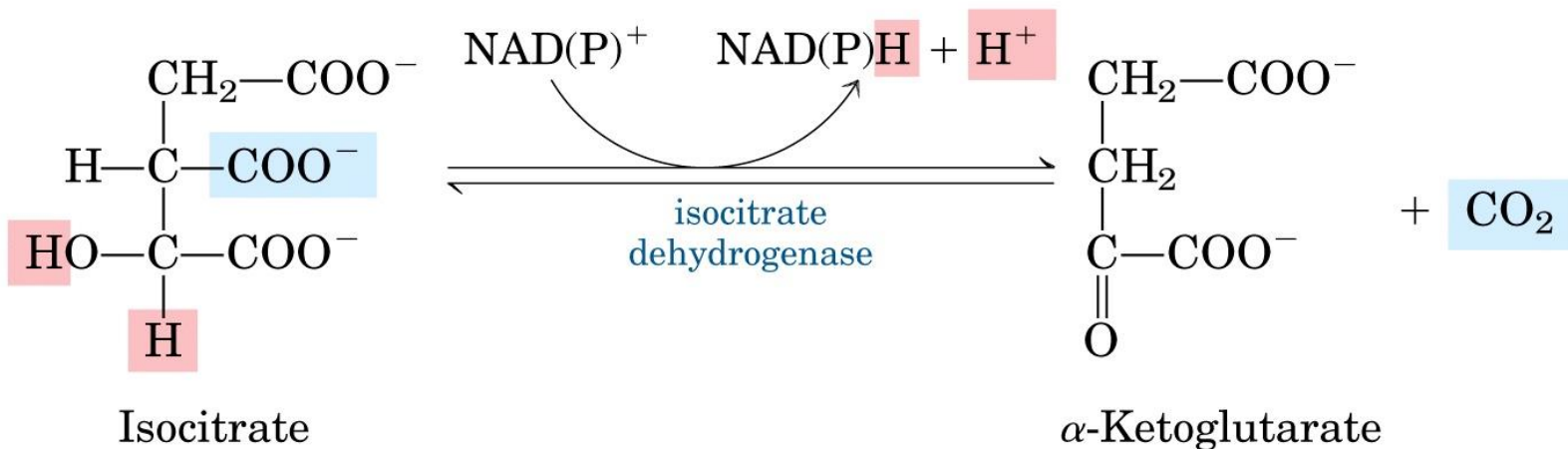
Results in interchange of H and OH. Changes structure and energy distribution within molecule. Makes easier for next enzyme to remove hydrogen



isocitrate → α -ketoglutarate

Isocitrate dehydrogenated and decarboxylated to give α -ketoglutarate

Stimulated by isocitrate, NAD⁺, ADP,
Inhibited by NADH and ATP



α -ketoglutarate \rightarrow succinyl CoA

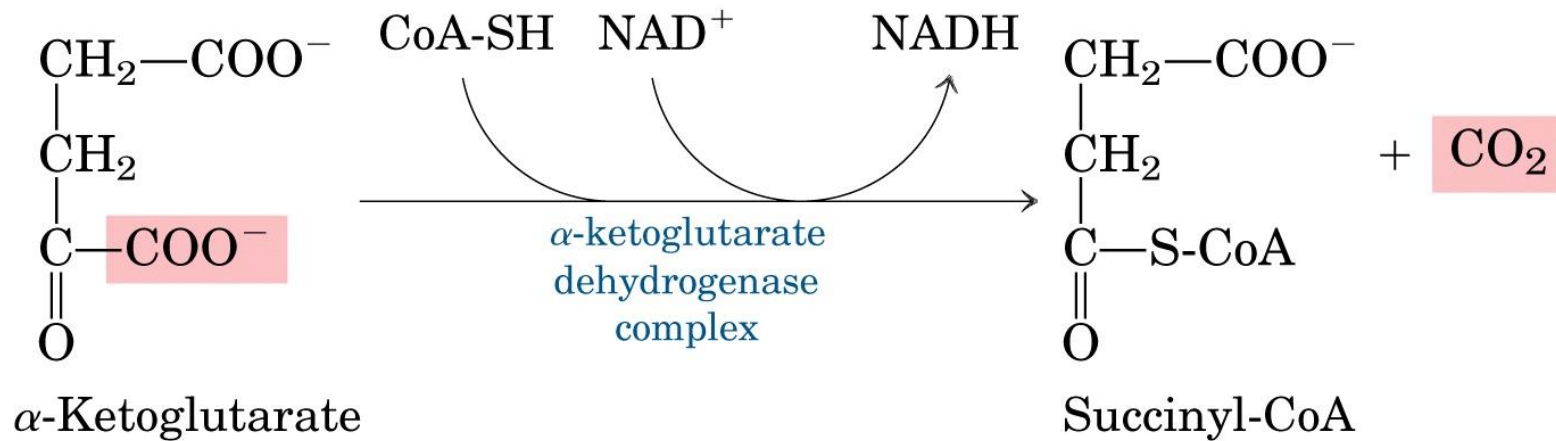
Series of reactions result in decarboxylation, dehydrogenation and incorporation of CoASH

Non-equilibrium reactions catalysed by α -ketoglutarate dehydrogenase complex

Results in formation of: $\text{CO}_2 + \text{NADH}$ and High energy bond

Stimulated by Ca^{2+}

Inhibited by NADH, ATP, Succinyl CoA

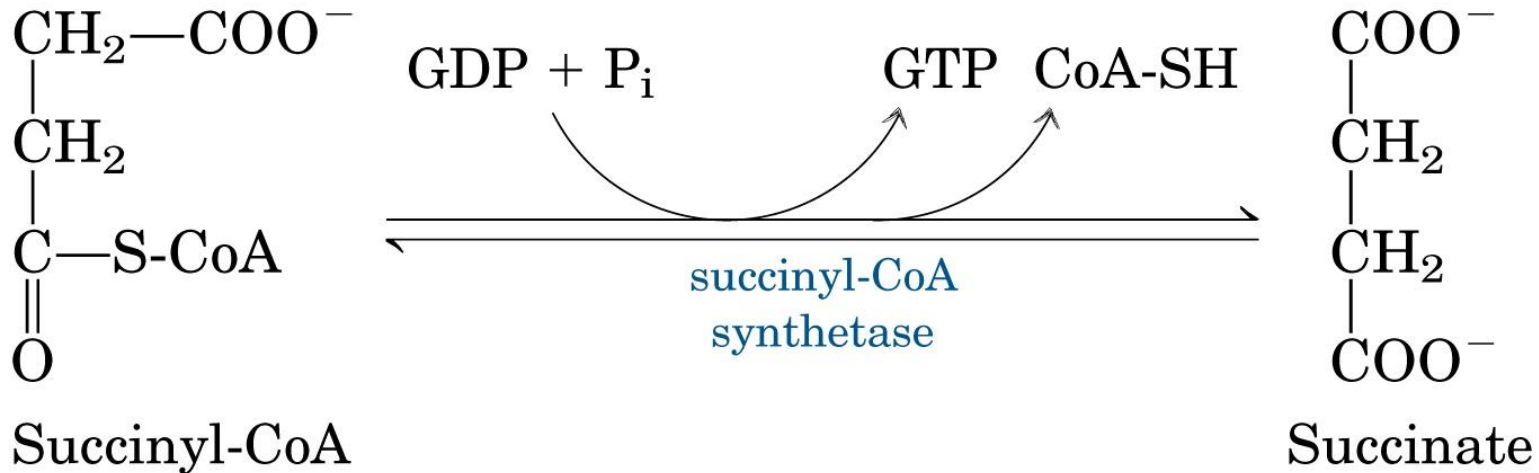


Succinyl CoA ↔ succinate

Equilibrium reaction, the succinyl thioester of CoA =energy rich bond.

Results in formation of: CoA-SH , GTP

GTP + ADP → GDP + ATP(nucleoside diphosphokinase)



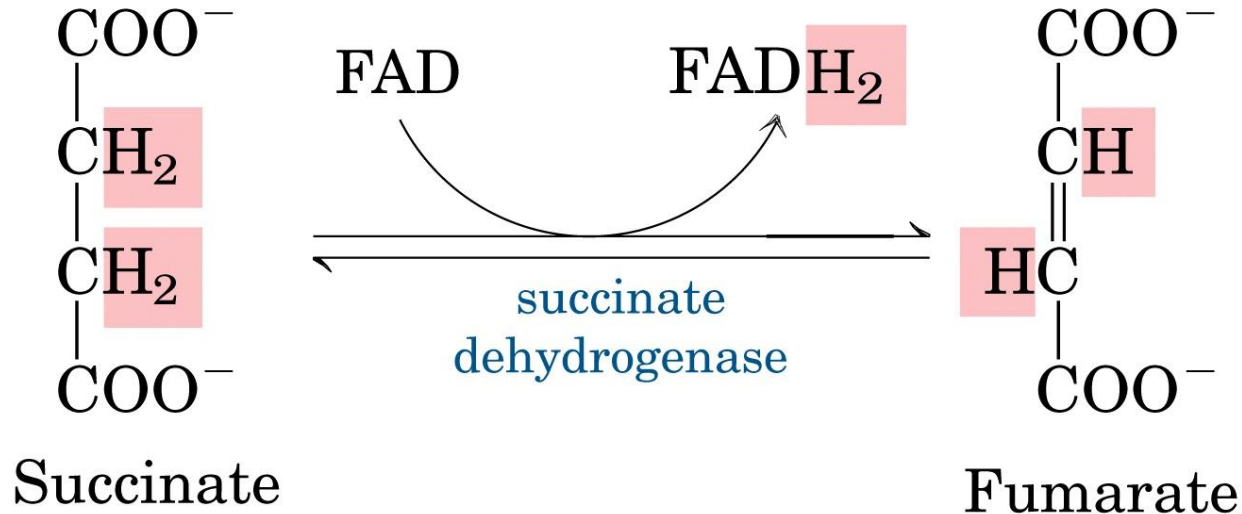
Succinate ↔ fumarate

Oxidation of succinate to fumarate reduces FAD to FADH₂

Succinate dehydrogenated to form fumarate

Equilibrium reaction catalysed by succinate dehydrogenase. Only
Krebs enzyme contained within inner mitochondrial membrane

Results in formation of FADH₂.



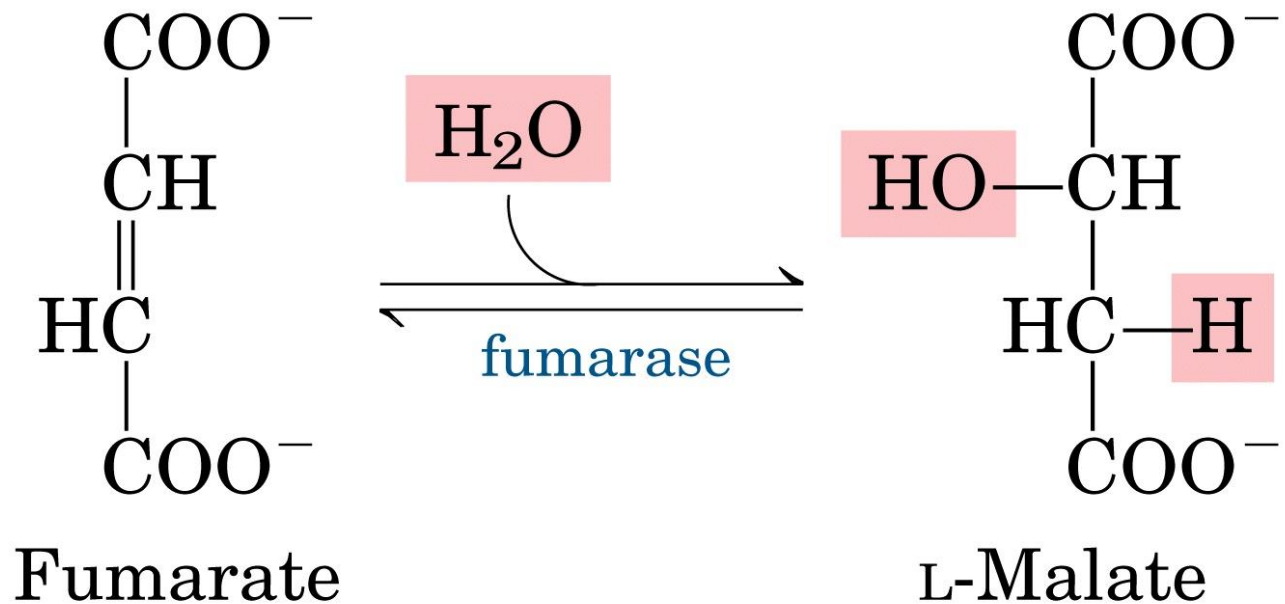
Fumarate ↔ malate

Addition of water to the double bond, to make the alcohol

Fumarate hydrated to form malate

Equilibrium reaction catalysed by fumarase

Results in redistribution of energy within molecule so next step can remove hydrogen



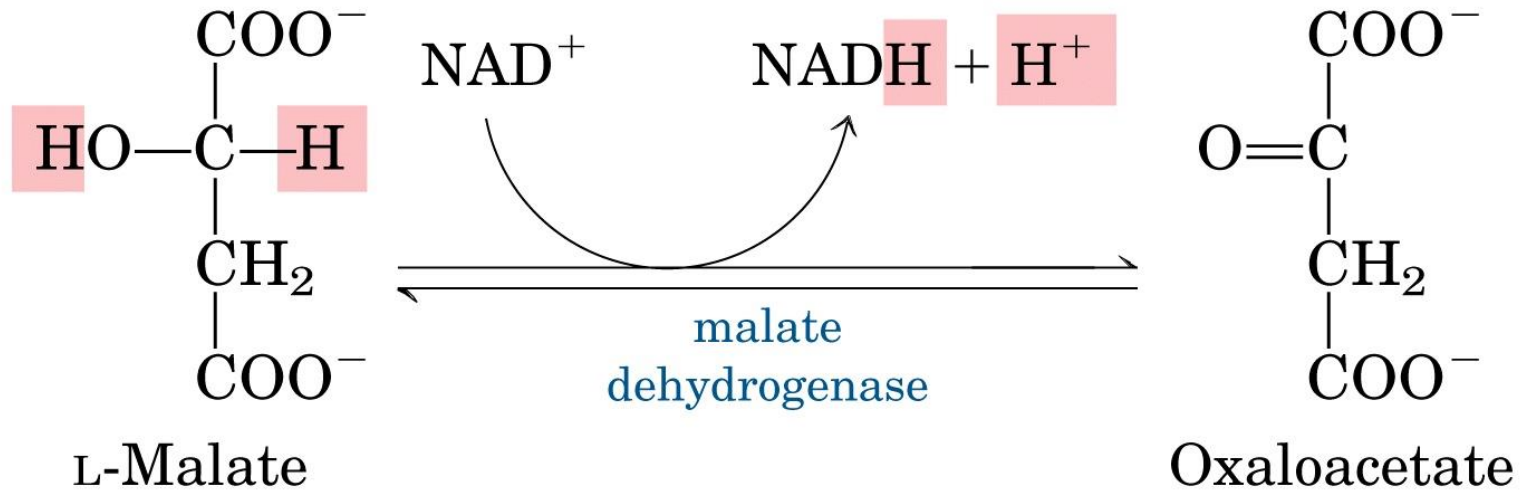
Malate ↔ oxaloacetate

Oxidation of Malate to Oxaloacetate reduces NAD^+ to NADH

Malate dehydrogenated to form oxaloacetate

Equilibrium reaction catalysed by malate dehydrogenase

Results in formation of $\text{NADH} + \text{H}^+$



Summary of TCA Cycle Reactions

acetyl CoA oxidized to 2CO_2

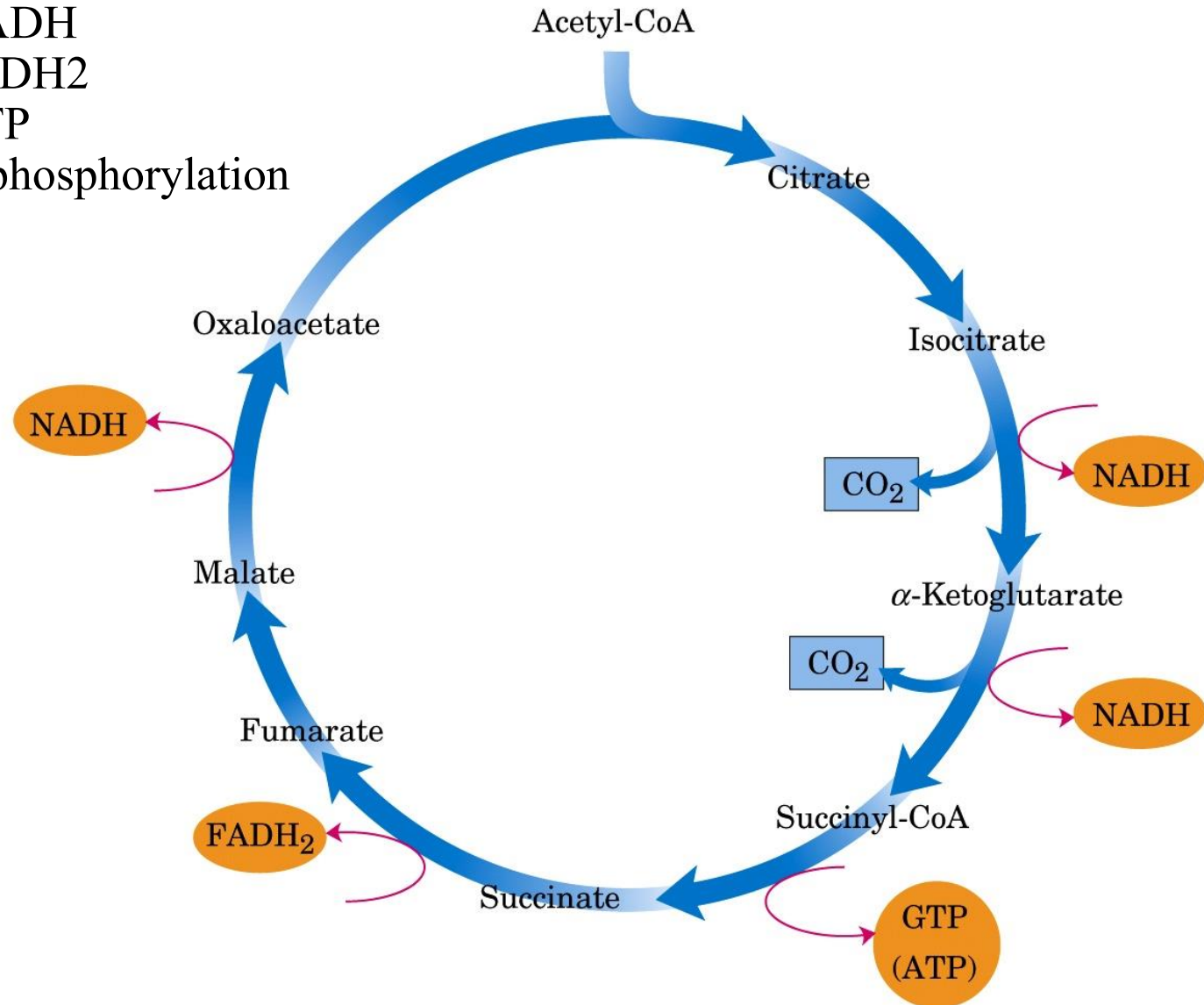
4 electron pairs

3NADH

1FADH₂

1GTP

Coupled to oxidative phosphorylation



All of the carbons that are input as **Pyruvate** are released as **CO₂**. This is as highly oxidized as carbon can get.

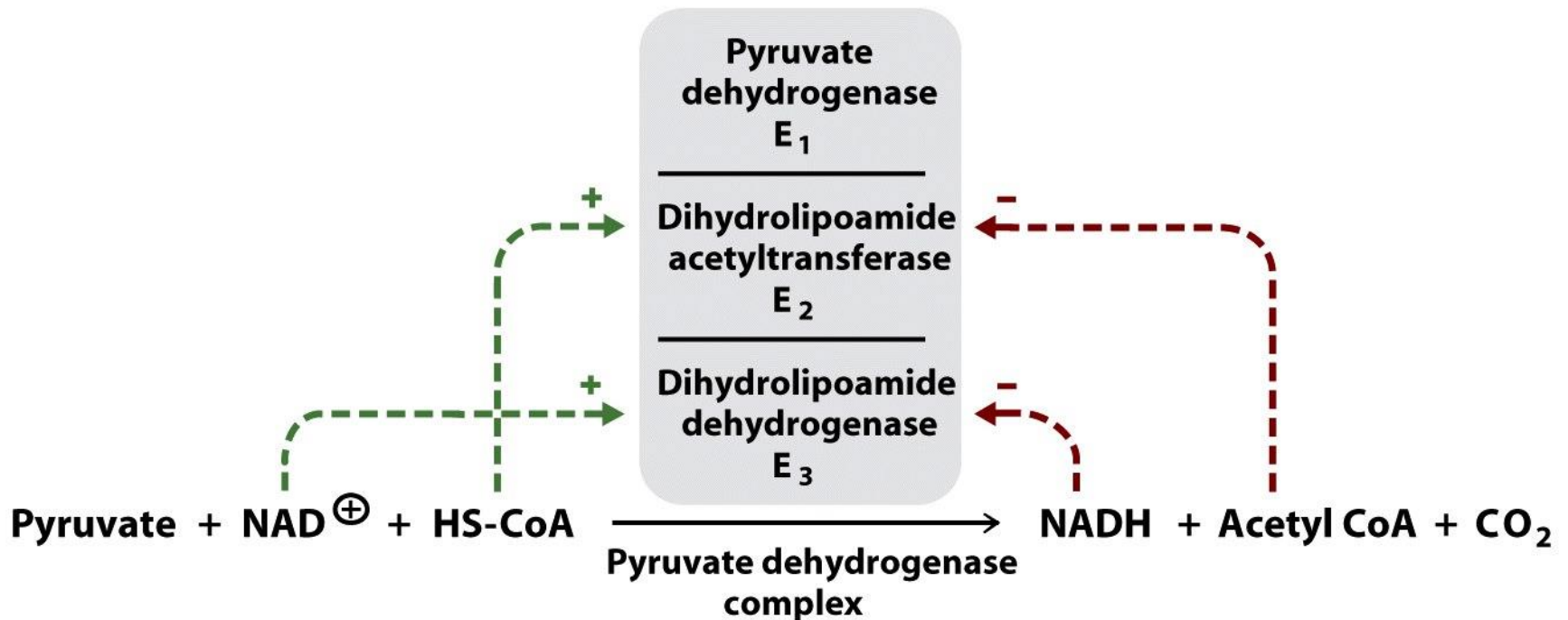
Oxidative decarboxylation: each time a CO₂ is produced one NADH is produced.

Locations of CO₂ release:

1. Pyruvate dehydrogenase: Pyruvate to Acetyl-CoA
2. Isocitrate dehydrogenase: Isocitrate to α-ketoglutarate
3. α-ketoglutarate dehydrogenase: α-ketoglutarate to succinyl-CoA

Regulation of PDH complex

- 1- inhibition by end products (NADH, Acetyl CoA). Increased levels of acetyl CoA and NADH inhibit E₂, E₃ in mammals and *E. coli*.
- 2- Feed back regulation by nucleotides (GTP)



3- Regulation by reversible phosphorylation phosphorylation/dephosphorylation of E₁

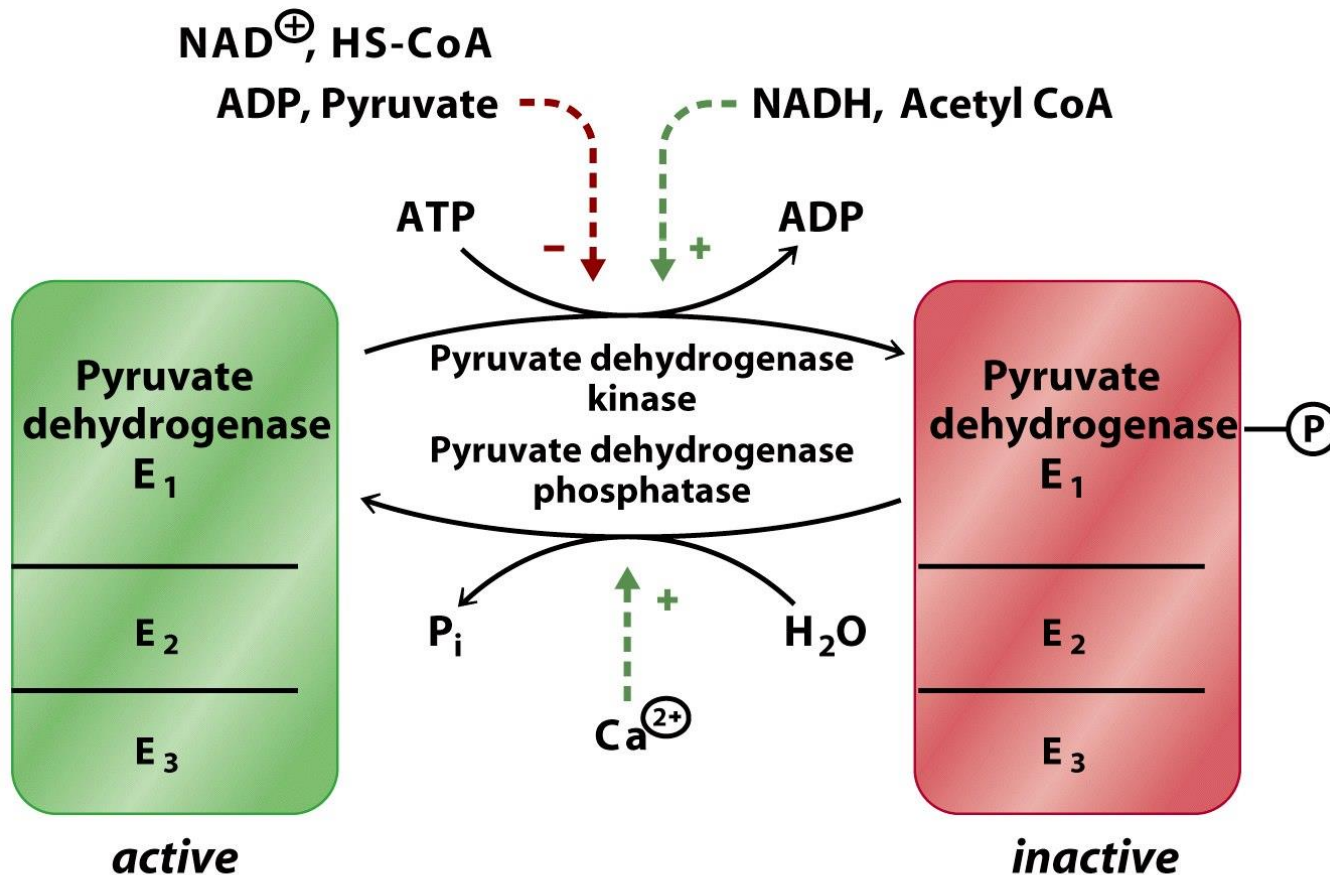


Figure 13-12 Principles of Biochemistry, 4/e
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Regulation of Krebs Cycle

- Cycle always proceeds in **same direction** due to presence of 3 non-equilibrium reactions catalysed by
 - **Citrate synthase**
 - NADH, ATP, citrate, succinyl CoA
 - + NAD⁺, ADP
 - **Isocitrate dehydrogenase**
 - NADH, ATP
 - + NAD⁺, ADP, Ca²⁺
 - **α-ketoglutarate dehydrogenase**
 - NADH, Succinyl CoA
 - + NAD⁺, Ca²⁺

Regulation of Krebs Cycle

- Flux through KC increases during exercise i.e as contractile activity increases
- 3 non-equilibrium enzymes inhibited by NADH
 - KC tightly coupled to ETC
 - If NADH decreases due to increased oxidation in ETC flux through KC increases
- Isocitrate dehydrogenase and α -ketoglutarate dehydrogenase also stimulated by Ca^{2+}

